# Application Note · PlasmaQuant MS



#### Challenge

Seawater matrices cause spectral and physical interferences from its high dissolved salt content (3.5% m/v)

### Solution

The aerosol dilution system of the PlasmaQuant MS allows for the direct analysis of seawater despite percent level of total dissolved solids (TDS) without prior sample dilution

# Intended audience

Relevant for environmental, research, aquaculture, mining, and more industries

# Multielement Determination of Trace Elements in Undiluted Seawater by ICP-MS

# Introduction

Nowadays, inductively coupled plasma mass spectrometry (ICP-MS) has become one of the most powerful analytical techniques for trace element analysis with high sensitivity, wide linear dynamic range, and quasi-simultaneous multielement detection capability.<sup>[1,2]</sup> However, having a principal disadvantage of polyatomic interferences combined with its weak tolerance for dissolved salts, makes it difficult for the ICP-MS to perform direct injection analysis of seawater due to high salt contents of approximately 3.5%. The high matrix load may possibly cause matrix effects, spectral interferences, clogging of the torch tip and the interface cone orifice. Furthermore, the concentrations of most trace metals in seawater are extremely low at ng/L levels.<sup>[1]</sup> To address these analytical challenges, a range of different sample preparation and introduction techniques have been used in combination with ICP-MS, such as solvent extraction,<sup>[3,4]</sup> coprecipitation,<sup>[5]</sup> and chelating resin adsorption.<sup>[6]</sup> However, these procedures are time-consuming and, in addition, could be a source of trace element contamination. Therefore, direct seawater analysis is favored using online aerosol dilution in combination with collision/reaction cell technology to provide efficient removal of spectral interferences.

This application note demonstrates the capability of the PlasmaQuant MS to handle undiluted seawater, profiting from its high plasma robustness and sensitivity.



# Materials and Methods

## Instrument settings

A PlasmaQuant MS ICP in combination with the CETAC ASX-560 autosampler and ESI injection valve was used for the analysis of 30 elements in four Seawater Certified Reference Materials from National Research Council Canada. The PlasmaQuant MS used is equipped with 1) aerosol dilution which allows for e.g. 10-fold dilution; 2) a unique interface including larger orifices preventing salt build-up and a 9 mm cone separation to reduce material deposition and blockage; 3) Nitrox option which offers the addition of  $N_2$  to the auxiliary gas to enhance the sensitivity of poorly ionized elements such as As and Se; 4) unique (patented) BOOST technology that allows a positive voltage (up to 10 V) on skimmer cone for sensitivity improvement in reaction mode; 5) a patented integrated Collision Reaction Cell (iCRC) to remove polyatomic species formed in the plasma and greatly improve the accuracy of an analysis.

All experiments were carried out in a routine analytical laboratory, and not under 'clean room' conditions. Instrument operating conditions are summarized in Table 1.

Parameter	Specification						
Plasma Gas Flow	10.5 L/min						
Auxiliary Gas Flow	1.50 L/min						
Sheath Gas Flow	0.83 L/min						
Nebulizer Gas Flow	0.27 L/min	0.27 L/min					
Sampling depth	7.0 mm						
Plasma RF Power	1.60 kW	1.60 kW					
Pump Rate	10 rpm – black/black PVC pump tubing (<1 mL/min)						
Stabilization delay	10 s						
iCRC Gas Setting	He 120 mL/min	<sup>11</sup> B and <sup>47</sup> Ti					
	He 180 mL/min	<sup>27</sup> Al, <sup>51</sup> V, <sup>52</sup> Cr, <sup>59</sup> Co, <sup>60</sup> Ni, and <sup>65</sup> Cu					
	No Gas	<sup>7</sup> Li, <sup>9</sup> Be, <sup>71</sup> Ga, <sup>85</sup> Rb, <sup>88</sup> Sr, <sup>98</sup> Mo, <sup>107</sup> Ag, <sup>114</sup> Cd, <sup>118</sup> Sn, <sup>121</sup> Sb, <sup>137</sup> Ba, <sup>139</sup> La, <sup>202</sup> Hg, <sup>205</sup> Tl, <sup>206+207+208</sup> Pb, <sup>209</sup> Bi, and <sup>238</sup> U					
	H <sub>2</sub> 180 mL/min	<sup>66</sup> Zn					
	H <sub>2</sub> 200 mL/min + Boost (4 V) + N <sub>2</sub> 40 mL/min	<sup>55</sup> Mn, <sup>56</sup> Fe, <sup>75</sup> As, and <sup>78</sup> Se					
Dwell Time	30 ms						
Scan per Replicate	10 (peak hopping, 1 pt/peak)						
No. of replicates	3						
Sample uptake time	0 s – OneFAST Sample Introduction system used						
Internal Standards	Sc, Y, In, Ir, and Tb, 50 μg/L, interpolation correction	on					

Table 1: Instrument settings – PlasmaQuant MS

# Samples and reagents

The following high purity reagents were used for all solution preparations:

- Deionized water (>18.2 MΩ/cm, Millipore Milli-Q)
- Nitric acid Supra-quality 69 % (ROTIPURAN<sup>®</sup> Supra)
- Sodium Chloride 99.999% trace metals basis (Sigma-Aldrich)

# Sample preparation

As part of the experiment, all undiluted CRMs were loaded onto the autosampler and the internal standards containing diluent were added online using a T-piece located directly after the exit port from the OneFast valve before the nebulizer. The final ratio of seawater to diluent was 1:20. Online dilution is recommended to minimize unwanted contamination. The CASS-5, CASS-6 and NASS-6 and NASS-7 Seawater Certified Reference Material for trace metals (National Research Council Canada, Ottawa, Ontario) were selected to evaluate accuracy. Spike recoveries of 1 and 10 ppb were also performed for most of the elements tested, except for some elements (e.g., Li, B and Sr) due to their relatively high concentration in the CRMs.

#### Calibration

A six-point external calibration was used for quantification with standard concentrations chosen relative to the expected concentration in the CRMs. Calibration solutions were prepared in 50 ml plastic tubes from SARSTEDT from

high-purity, single element solutions (CertiPUR® 1000 mg/L) in 1% (v/v)  $\rm HNO_3$  and 35 g/L of NaCl covering the following concentration ranges:

Table 2: Concentration Ranges

Element	Concentration range (µg/L)		
Pb	0.01 - 2		
Cd	0.02 - 2		
Нд	0.1 - 2		
Be, Co, Ni, Cu, Ga, Rb, Mo, Ag, Sn, Sb, Ba, La, Tl, Bi and U	0.1 - 20		
for Cr, Mn, Fe and Zn	0.2 - 20		
Al, Ti, V, As and Se	1 - 20		
Li and Sr	1 - 200		
В	10 - 200		

The calibration graphs in Figure 1 show good examples of the performance from the PlasmaQuant MS at defined concentration levels. All calibrations were measured with correlation coefficients better than 0.9997.

#### Measurement

According to the expected interferences on certain mass/charge ratios, different isotopes were measured using collision gas (He) or a reaction gas (H<sub>2</sub>) of the iCRC. Besides the two available gases, PlasmaQuant MS allows the users to recover analytes signals after reaction took place on iCRC through the Boost patented technology applied on skimmer cone.

Due to low ionization efficiency of some elements, such as As and Se, the Nitrox option allows for the addition of Nitrogen directly to the plasma through the auxiliary port which improves the ionization 3 - 4 x for As and Se.

To ensure the highest precision and accuracy in our measurements, we selected five different condition sets for each measurement. These condition sets involved varying gas flows, including Helium at 120 and 180 mL/min, No Gas, Hydrogen at 180 and 200 mL/min combined with Boost (4 V), and Nitrogen at 40 mL/min. Each condition set was associated with specific isotopes, as detailed in Table 3.

During the experiments, all condition sets were run with switching times set at 10 seconds. For data recording purposes, we calculated three average values from ten scans each. These average values, along with their standard deviations, were used in the final calculations.

The total sampling time, including rinse and sample uptake delays, was about three minutes per sample.

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Table 3: Expected	interferences and	used iCRC gases
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lsotope	Expected interference	iCRC mode	Isotope	Expected interference	iCRC mode
<sup>7</sup> Li	-	No Gas	<sup>78</sup> Se	<sup>40</sup> Ar <sup>38</sup> Ar	H <sub>2</sub>
9Be	-	No Gas	<sup>85</sup> Rb	-	No Gas
<sup>11</sup> B	-	He (due to high concentration)	<sup>88</sup> Sr	-	No Gas
<sup>27</sup> AI	<sup>11</sup> B <sup>16</sup> O	Не	<sup>98</sup> Mo	-	No Gas
<sup>47</sup> Ti	<sup>31</sup> P <sup>16</sup> O	Не	<sup>107</sup> Ag	-	No Gas
<sup>51</sup> V	<sup>35</sup> Cl <sup>16</sup> O	Не	<sup>114</sup> Cd	<sup>98</sup> Mo <sup>16</sup> O	No Gas
<sup>52</sup> Cr	<sup>40</sup> Ar <sup>12</sup> C; <sup>35</sup> Cl <sup>16</sup> O <sup>1</sup> H	Не	<sup>118</sup> Sn	-	No Gas
⁵⁵Mn	<sup>39</sup> K <sup>16</sup> O	H <sub>2</sub>	<sup>121</sup> Sb	-	No Gas
<sup>56</sup> Fe	<sup>40</sup> Ar <sup>16</sup> O; <sup>40</sup> Ca <sup>16</sup> O	H <sub>2</sub>	<sup>137</sup> Ba	-	No Gas
<sup>59</sup> Co	<sup>43</sup> Ca <sup>16</sup> O; <sup>42</sup> Ca <sup>16</sup> O <sup>1</sup> H; <sup>24</sup> Mg <sup>35</sup> Cl	Не	<sup>139</sup> La	-	No Gas
<sup>60</sup> Ni	<sup>44</sup> Ca <sup>16</sup> O; <sup>25</sup> Mg <sup>35</sup> Cl; <sup>23</sup> Na <sup>37</sup> Cl	Не	<sup>202</sup> Hg	-	No Gas
<sup>65</sup> Cu	<sup>49</sup> Ti <sup>16</sup> O; <sup>25</sup> Mg <sup>40</sup> Ar	Не	<sup>205</sup> TI	-	No Gas
<sup>66</sup> Zn	<sup>26</sup> Mg <sup>40</sup> Ar; <sup>31</sup> P <sup>35</sup> Cl	H <sub>2</sub>	<sup>206+207+208</sup> Pb	-	No Gas
<sup>71</sup> Ga	-	No Gas	<sup>209</sup> Bi	-	No Gas
<sup>75</sup> As	<sup>40</sup> Ar <sup>35</sup> Cl; <sup>40</sup> Ca <sup>35</sup> Cl	H <sub>2</sub>	<sup>238</sup> U	-	No Gas

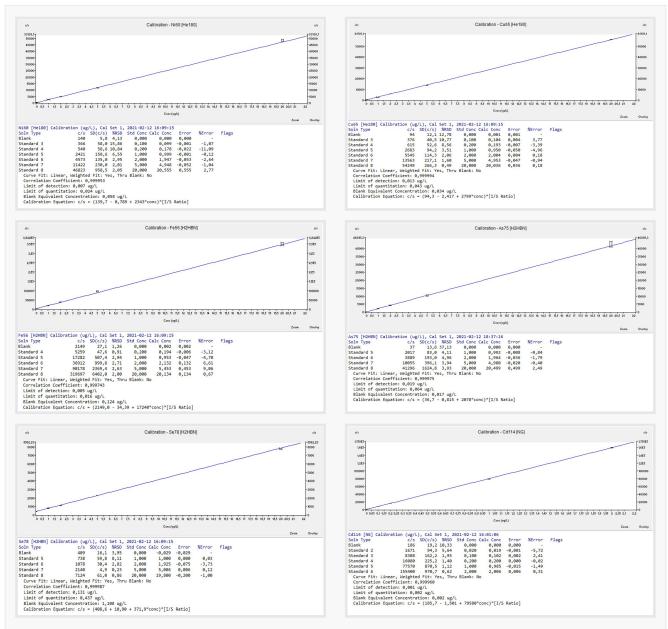


Figure 1: Calibration graphs of some of the most challenging elements investigated (e.g., <sup>60</sup>Ni, <sup>65</sup>Cu, <sup>56</sup>Fe, <sup>75</sup>As, <sup>78</sup>Se, and <sup>114</sup>Cd)

# Results and Discussion

# The obtained results for Seawater CRMs samples are given in Table 4, 5, 6 and 7.

lsotope	mean	%RSD	Certified	Unc.	%REC	1 ppb Spike REC., %	10 ppb Spike REC., %
<sup>7</sup> Li	162	0.7	-	-	-	-	-
9Be	<0.0003	-	-	-	-	94	104
<sup>11</sup> B	4153	1.7	-	-	-	-	-
<sup>27</sup> AI	0.26	38.1	-	-	-	109	99
<sup>47</sup> Ti	0.11	-	-	-	-	-	-
<sup>51</sup> V	1.28	3.1	1.32	0.14	97	88	113
<sup>52</sup> Cr	0.098	20.9	0.106	0.013	92	100	97
⁵⁵Mn	2.64	6.1	2.62	0.2	101	99	97
<sup>56</sup> Fe	1.58	6.1	1.44	0.11	110	86	86
<sup>59</sup> Co	0.111	7.6	0.095	-	117	99	102
<sup>60</sup> Ni	0.32	8.4	0.33	0.023	97	101	112
<sup>65</sup> Cu	0.37	2.1	0.38	0.028	97	97	113
<sup>66</sup> Zn	0.70	8.3	0.719	0.068	97	96	122
<sup>71</sup> Ga	0.03	12.9	-	-	-	96	96
<sup>75</sup> As	1.37	4.4	1.24	0.09	110	100	96
<sup>78</sup> Se	0.18	3.0	-	-	-	92	93
<sup>85</sup> Rb	96.4	0.1	-	-	-	93	96
<sup>88</sup> Sr	6425	0.6	-	-	-	-	-
<sup>98</sup> Mo	10.0	0.4	9.82	0.72	102	110	104
<sup>107</sup> Ag	<0.002	-	-	-	-	91	100
<sup>114</sup> Cd	0.025	2.8	0.0215	0.0018	116	95	96
<sup>118</sup> Sn	<0.08	-	-	-	-	86	94
<sup>121</sup> Sb	0.38	3.9	-	-	-	91	101
<sup>137</sup> Ba	7.02	1.5	-	-	-	101	104
<sup>139</sup> La	0.010	0.0	-	-	-	93	102
<sup>202</sup> Hg	0.012	8.2	-	-	-	82	92
<sup>205</sup> TI	0.012	0.0	-	-	-	92	99
<sup>206+207+208</sup> Pb	0.013	0.0	0.011	0.002	114	89	97
<sup>209</sup> Bi	<0.001	-	-	-	-	90	99
<sup>238</sup> U	3.03	1.4	3.18	0.11	95	91	100

Table 4: Concentration results (in  $\mu$ g/L) of CASS-5, precision, and accuracy (n=3; n.d. not determined)

lsotope	mean	%RSD	Certified	Unc.	%REC	1 ppb Spike REC., %	10 ppb Spike REC., %
<sup>7</sup> Li	n.d.	-	-	-	-	-	-
<sup>9</sup> Be	n.d.	-	-	-	-	-	-
<sup>11</sup> B	4102	1.2	4090	100	100	-	-
<sup>27</sup> AI	n.d.	-	-	-	-	-	-
<sup>47</sup> Ti	n.d.	-	-	-	-	-	-
<sup>51</sup> V	0.53	-	0.5	0.12	106	99	106
<sup>52</sup> Cr	0.12	-	0.1	0.016	120	101	116
⁵⁵Mn	2.19	2.7	2.22	0.12	99	98	97
<sup>56</sup> Fe	1.48	8.1	1.56	0.12	95	103	110
<sup>59</sup> Co	0.07	10.0	0.067	0.005	105	99	97
<sup>60</sup> Ni	0.52	0.4	0.418	0.04	124	102	98
<sup>65</sup> Cu	0.57	3.5	0.53	0.032	108	95	95
<sup>66</sup> Zn	1.27	1.6	1.27	0.18	100	104	91
<sup>71</sup> Ga	n.d.	-	-	-	-	-	-
<sup>75</sup> As	1.32	-	1.04	0.1	127	101	86
<sup>78</sup> Se	2.91	-	-	-	-	100	101
<sup>85</sup> Rb	n.d.	-	-	-	-	-	-
<sup>88</sup> Sr	n.d.	-	-	-	-	-	-
<sup>98</sup> Mo	9.15	0.8	9.15	0.52	-	-	-
<sup>107</sup> Ag	n.d.	-	-	-	-	-	-
<sup>114</sup> Cd	0.020	10.0	0.022	0.002	91	99	99
<sup>118</sup> Sn	n.d.	-	-	-	-	-	-
<sup>121</sup> Sb	n.d.	-	-	-	-	-	-
<sup>137</sup> Ba	7.95	-	-	-	-	102	103
<sup>139</sup> La	n.d.	-	-	-	-	-	-
<sup>202</sup> Hg	n.d.	-	-	-	-	-	-
<sup>205</sup> TI	n.d.	-	-	-	-	-	-
<sup>206+207+208</sup> Pb	0.01	0.0	0.011	0.004	91	99	105
<sup>209</sup> Bi	n.d.	-	-	-	-	-	-
<sup>238</sup> U	2.60	4.6	2.92	0.42		-	-

# Table 5: Concentration results (in $\mu$ g/L) of CASS-6, precision, and accuracy (n=3; n.d. not determined)

Isotope	mean	%RSD	Certified	Unc.	%REC	1 ppb Spike REC., %	10 ppb Spike REC., %
<sup>7</sup> Li	166	0.2	-	-	-	-	-
<sup>9</sup> Be	<0.0003	-	-	-	-	98	96
<sup>11</sup> B	4262	0.5	-	-	-	-	-
<sup>27</sup> AI	<0.04	-	-	-	-	-	91
<sup>47</sup> Ti	0.08	12.0	-	-	-	-	-
<sup>51</sup> V	1.49	10.0	1.46	0.17	102	87	99
<sup>52</sup> Cr	0.12	-	0.118	0.008	102	103	100
⁵⁵Mn	0.52	0.7	0.53	0.05	98	99	95
<sup>56</sup> Fe	1.19	1.9	0.495	0.046	241**	114	88
<sup>59</sup> Co	0.014	2.0	0.015	-	93	99	99
<sup>60</sup> Ni	0.29	2.5	0.301	0.025	98	109	99
<sup>65</sup> Cu	0.242	2.3	0.248	0.025	98	81	95
<sup>66</sup> Zn	<0.008	-	0.257	0.02	-	117	92
<sup>71</sup> Ga	0.033	1.7	-	-	-	97	97
<sup>75</sup> As	1.71	3.6	1.43	0.12	119	113	97
<sup>78</sup> Se	<0.13	-	-	-		-	107
<sup>85</sup> Rb	96.8	0.5	-	-	-	-	92
<sup>88</sup> Sr	6440	0.2	-	-	-	-	-
<sup>98</sup> Mo	9.35	0.4	9.89	0.72	95	105	93
<sup>107</sup> Ag	<0.002	-	-	-	-	97	87
<sup>114</sup> Cd	0.036	7.9	0.0311	0.0019	116	96	96
<sup>118</sup> Sn	<0.002	-	-	-	-	76	78
<sup>121</sup> Sb	0.25	2.2	-	-	-	94	93
<sup>137</sup> Ba	6.56	0.4	-	-	-	97	92
<sup>139</sup> La	0.008	9.4	-	-	-	97	97
<sup>202</sup> Hg	<0.005	-	-	-	-	99	93
<sup>205</sup> TI	0.004	2.0	-	-	-	95	97
<sup>206+207+208</sup> Pb	<0.0003	-	0.006	0.002	-	97	99
<sup>209</sup> Bi	<0.001	-	-	-	-	98	99
<sup>238</sup> U	3.00	0.4	3	-	100	101	99

# Table 6: Concentration results (in $\mu$ g/L) of NASS-6, precision, and accuracy (n=3; n.d. not determined)

\*\* Fe contaminated at 0.5 ppb level due to a contamination of the sample

Isotope	mean	%RSD	Certified	Unc.	%REC	1 ppb Spike REC., %	10 ppb Spike REC., %
<sup>7</sup> Li	153	0.0	-	-	-	-	-
<sup>9</sup> Be	<0.0003	-	-	-	-	98	104
<sup>11</sup> B	3718	1.9	3750	120	99	-	-
<sup>27</sup> AI	0.36	5.0	-	-	-	80	107
<sup>47</sup> Ti	<0.05	-	-	-	-	-	-
<sup>51</sup> V	1.34	-	1.3	0.08	103	-	111
<sup>52</sup> Cr	n.d.	-	0.107	0.016	-	-	-
⁵⁵Mn	0.74	3.4	0.75	0.06	99	105	95
<sup>56</sup> Fe	0.37	1.9	0.351	0.026	106	105	96
<sup>59</sup> Co	0.020	3.2	0.0146	0.0014	133**	102	100
<sup>60</sup> Ni	0.28	1.4	0.248	0.018	113	105	99
<sup>65</sup> Cu	0.19	0.4	0.199	0.014	93	105	100
<sup>66</sup> Zn	0.42	1.5	0.42	0.08	100	112	111
<sup>71</sup> Ga	0.06	1.3	-	-	-	101	102
<sup>75</sup> As	1.34	7.2	1.26	0.06	107	95	80
<sup>78</sup> Se	<0.13	-	-	-	-	97	93
<sup>85</sup> Rb	98.0	0.1	-	-	-	-	109
<sup>88</sup> Sr	6459	0.2	-	-	-	-	-
<sup>98</sup> Mo	9.7	0.5	9.29	0.4	104	95	101
<sup>107</sup> Ag	<0.002	-	-	-	-	93	103
<sup>114</sup> Cd	0.016	17.7	0.0161	0.0016	100	101	103
<sup>118</sup> Sn	<0.08	-	-	-	-	90	96
<sup>121</sup> Sb	0.17	2.6	-	-	-	92	102
<sup>137</sup> Ba	4.74	0.9	-	-	-	97	104
<sup>139</sup> La	0.007	10.9	0.006	-	108	94	101
<sup>202</sup> Hg	0.63	-	-	-	-	84	102
<sup>205</sup> TI	0.011	6.7	-	-	-	92	99
<sup>206+207+208</sup> Pb	0.004	-	0.0026	0.0008	133***	98	104
<sup>209</sup> Bi	0.002	-	-	-	-	91	96
<sup>238</sup> U	2.86	0.5	2.87	0.16	100	93	100

# Table 7: Concentration results (in $\mu$ g/L) of NASS-7, precision, and accuracy (n=3; n.d. not determined)

\*\* Co contaminated at 5 ppt; \*\*\* Pb contaminated 0.5 ppt level due to a contamination of the sample

# Accuracy

Certified reference materials were used to evaluate accuracy besides spike recovery efficiency. The mean results are in good agreement with the expected or assigned values provided. All certified elements were within 80-120% recovery range (Figure 2). It should be mentioned that such accuracy can only be achieved with sensitive and precise measurement.

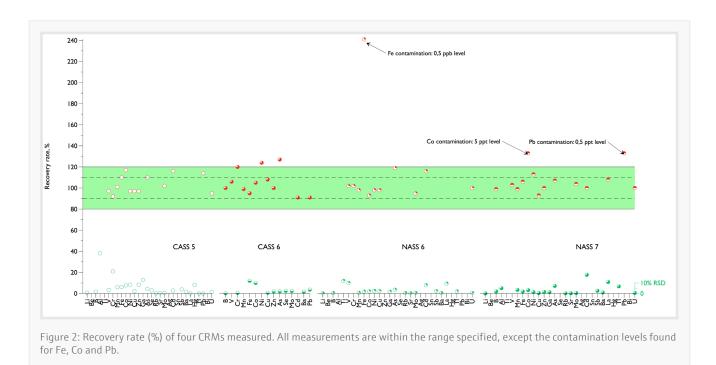
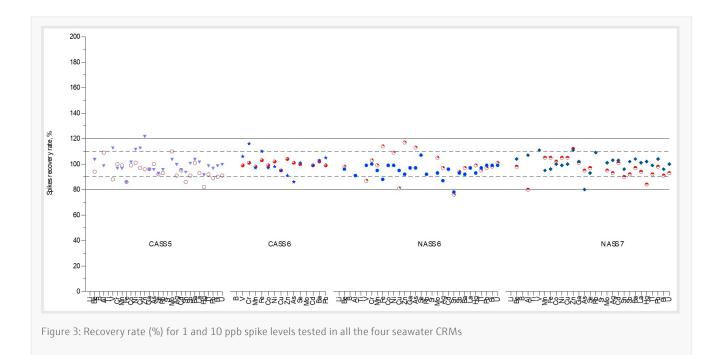


Figure 3 shows excellent spike recoveries for all elements in the certified reference materials at both spike levels (1 and 10 ppb). Most spike recoveries are within 90-110% range.



# Precision

The relative standard deviation (RSD) was used as a parameter to assess the precision of the measurements (Figure 2). On average RSD of 4.2% was achieved. Besides the fact, that RSDs are near or higher than 10%, mainly due to contamination level in the ppt concentration range, all the results with the range of ppt to ppb showed great precision and accuracy with the methodology applied during all the measurements.

# **Detection limits**

Table 7 provides two types of detection limits:

- Detection Limits (DLs): These correspond to three times the standard deviation of the calibration blank, calculated based on three measurements (n=3).
- Method Detection Limits (MDLs): These were determined by taking ten measurements of blank seawater (composed of 35 g/L NaCl with 1% (v/v)  $HNO_3$ ) at different points during the experiment. The MDLs were then calculated by multiplying the standard deviation of these ten measurements by three.

Although the MDLs are mostly single-digit ppt levels, some MDLs could be improved by using high-purity NaCl or artificial seawater.

lsotope	DL	MDL	Isotope	DL	MDL
<sup>7</sup> Li	2	14	<sup>78</sup> Se	131	38
<sup>9</sup> Be	0.3	2	<sup>85</sup> Rb	6	4
<sup>11</sup> B	116	1199	<sup>88</sup> Sr	7	5
<sup>27</sup> AI	43	205	<sup>98</sup> Mo	0.3	10
<sup>47</sup> Ti	51	21	<sup>107</sup> Ag	2	4
<sup>51</sup> V	115	15	<sup>114</sup> Cd	1	2
<sup>52</sup> Cr	14	11	<sup>118</sup> Sn	2	2
⁵⁵Mn	17	3	<sup>121</sup> Sb	1	3
<sup>56</sup> Fe	5	15	<sup>137</sup> Ba	2	2
<sup>59</sup> Co	2	2	<sup>139</sup> La	0.1	2
<sup>60</sup> Ni	7	51	<sup>202</sup> Hg	5	11
<sup>65</sup> Cu	13	29	<sup>205</sup> TI	0.1	3
<sup>66</sup> Zn	8	63	<sup>206+207+208</sup> Pb	0.3	2
<sup>71</sup> Ga	2	2	<sup>209</sup> Bi	1	2
<sup>75</sup> As	19	4	238U	0.1	0.1

Table 8: Detection Limits and Method detection limits for all the measured elements; Values in ng/L

# **Internal Standard Stability**

Seawater matrices create several impacts on ICP-MS, among them, sample uptake, nebulization, and aerosol transport, as well ionization suppression on plasma are the most common ones. The superior robustness of the PlasmaQuant MS interface combined with aerosol dilution and ESI switching valve allows high solids matrices to be aspirated, such as Seawaters for a prolonged period of time without any significant drift or loss of sensitivity which is currently observed in this kind of matrices. Such robustness extends maintenance time on the cones, better performance of Control Checks without compromising DLs over time.

Figure 4 shows the Internal standards stability remained within the  $\pm 20\%$  control limits in all condition's modes used during 1hour and 20 minutes run of PlasmaQuant MS while analyzing 25 undiluted seawater CRMs samples including blanks and calibration standards.

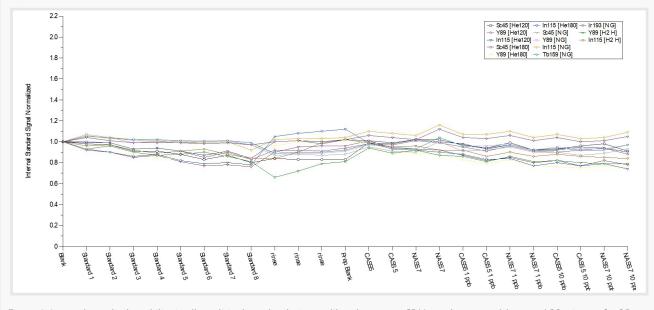


Figure 4: Internal standards stability in all condition's modes during undiluted seawater CRMs analyzes over 1 hour and 20 minutes for 25 samples

# Summary

The PlasmaQuant MS with aerosol dilution system allows for the direct analysis of undiluted seawater or any high matrix samples above 0.2% TDS limit frequently established for ICP-MS. Time consuming sample preparation and potential contamination due to samples liquid dilution are significantly reduced in combination with ESI injection valve which is an effective way of measuring ultra-trace concentration in high matrices.

Worldwide, the use of an accurate method is invaluable for environmental laboratories that require high-throughput seawater analysis capabilities.

The unique PlasmaQuant MS interface with larger orifices prevents salt build-up, reduces material deposition and blockage which reduces instrument downtime which is also important for high-throughput labs.



Figure 5: PlasmaQuant MS

### **Recommended device configuration**

Table 9: Overview of devices, accessories, and consumables

Article	Article number	Description
PlasmaQuant MS - highly sensitive, robust, and reliable ICP-MS Instrument	818-08010-2	PlasmaQuant MS is optimized for the analysis of tough samples such as seawater, soil, sludge, and combustibles
Cetac ASXPress Plus for PQMS	810-88017-0	The ASXpress plus Rapid Sample Introduction Accessory reduces time required for autosampler movement, sample uptake, stabilization, and rinse operations, thereby reducing sample run times significantly
Nitrox upgrade Kit for PQMS and PQMS Elite instruments	810-88025-0	Nitrox is vital in petrochemical processes, enabling carbon removal with oxygen. It's also advised for applications demanding low arsenic or selenium detection limits, enhancing sensitivity through nitrogen addition

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